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Study of Molecular Dynamics of a Nematic Main Chain Liquid Crystalline Polyester by Dielectric Spectroscopy

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STUDY OF MOLECULAR DYNAMICS OF A NEMATIC MAIN CHAIN LIQUID CRYSTALLINE POLYESTER BY DIELECTRIC SPECTROSCOPY

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Dielectric spectroscopy was applied to study the molecular dynamics of a nematic main-chain liquid polyester using crystalline а Du Pont Dielectric Analyser. The dielectric measurements frequency range from 0.03 to 100000Hz to 160°C leading from -100 temperature range from transition region to the up the isotropic region. Dielectricrelaxation processes α, β, γ and δ observed and analysed. activation Apparent the processes were calculated. energies of

INTRODUCTION

Dielectric relaxation spectroscopy is a useful and applied method for estimating the nature and extend of cooperative motion of dipoles, their relaxation frequencies at different temperatures 1-8. The dipoles associated with sites on the polymer chain are of three kinds: parallel, perpendicular to the main chain and side chain they dipoles. Major location for are) permanent electric dipoles in LCP's mesogenic are groups having effective dipole moment. The length of is spacer influential in determining the mesophase behaviour. dielectric activity of the spacer arises from its flexibility allowing reorientations of mesogenes.

The polymer backbone frequently carries polar groups and may be dielectrically active. Liquid crystalline polymers show dielectric phenomena familiar from low molecular crystals but the polymer chains present considerable impediment to molecular motions with radically changed dynamics. The molecular dynamics are determined by transition between different conformations of the polymer backbone. In case of LCP's α, β, γ and δ relaxations are recognized. The α and β processes are assumed associated with the motion of the transverse component of the dipole moment (β -oscillating local motions, α -segment around backbone), δ process is affected by the rotation longitudinal component of dipole moment, the 7 nematic is connected with motion within the spacer. process

EXPERIMENTAL

order to carry out a detailed analysis of the molecular response in the main chain liquid crystalline polyester, the dielectric measurements were made analyzed over а frequency from 0.03 to 100000Hz and from -100 to 160°C, temperature interval ranging from 7 relaxation region to the isotropisation. Dielectric measurements were performed with a Du Pont 2970 Dielectric The circular samples of the polymer diameter 0.025m and thickness 60µm ware prepared in chloroform and vacum dried. They were between two parallel gold plate sensors. The measurements controlled. were fully computer

The nematic liquid crystalline polyester (called DDA-9) poly(2,2 dimethyloazoxy benzene dodecane dioyl):

$$-[-0] = N = N = 0$$
 $CH_3 = CH_3$
 $CH_3 = CH_3$
 $CH_3 = 0$
 CH_2
 $CH_3 = 0$
 CH_3
 C

was prepared and characterized as previously described. A systematic investigation of the influence of chain length on the level of order for DDA-9 was investigated in the previous work. In the oligomeric range aliphatic chain ends destabilize the liquid crystalline (LC) order.

The order parameter of the mesogen increases strongly with increasing chain length and levels off at DP 10 displaying values larger then low molecular liquid crystal (LMLC). Molecular mass dependence of the isotropization temperature has been established for the unfractionated polymer samples and for fractions¹⁰. The polymer DDA-9 (M_n=18000) is characterized by a crystalline-nematic transition at 118°C and a nematic-isotropic transition at 163°C (as measured by DSC¹¹).

RESULTS AND DISCUSSION

interaction between the applied electric field the dipole moment of molecules changes the distribution of their reorientation. Dielectric properties of depend on both frequency ω and temperature T. Four major dielectric reported in are permittivity ϵ' and loss factor ϵ'' both provide valuable (ε' measures about the molecular motion information ε′′ represents dipoles, the alignment of dissipation dipoles and move ions), required to align and ionic conductivity, which becomes factor $tg\delta = \epsilon''/\epsilon'$ significant in the liquid state $\sigma=\epsilon''\omega\epsilon$. A survey of the absorption ϵ' , ϵ'' and $tg\delta$ as a function of dielectric frequency and temperature was made.

The temperature dependence of $lg(tg\delta)$ obtained for different frequencies is given in Fig.1.

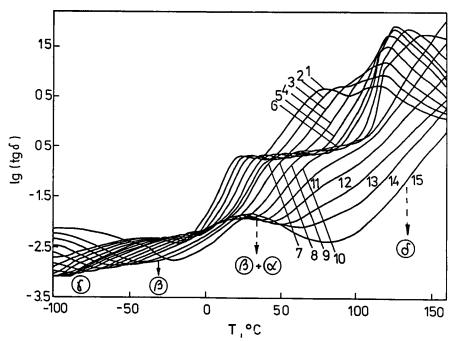


FIGURE 1 Lg(tg δ) versus temperature T measured for 15 frequencies(0.03 - 100000Hz, curves 1-15) γ, β, α and δ relaxation processes are visible.

The following four dielectric dispersions (dominated by involving the ester-group of the polymer) range -100 to 160° C: the γ observed in the temperature high relaxation process occurs at frequency at temperatures close to -90° C, the β relaxation process in of -40°C temperature frequency and at the secondary β relaxation at higher frequency and temperature in the vicinity of the polymer Tg=10⁷C, finally α transition temperature associated with reorientation at Tq involving mesogenic groups and polar ester groups the alkyl spacer in a rotational the polymer backbone. The a process segments around hundreds times slower (low frequency) than the β secondary process of the oscillating motion of the same dipoles.

Figure 2 shows the curves of ϵ'' and ϵ' plotted adjust temperature for the lowest and the highest applied frequencies (0.03 and 30000 Hz.) in order to illustrate the presence of primary β' and secondary β'' relaxation at regions of low (-40°C) and high (20°C) temperatures respectively (curves 2 and 3 on figure 3).

Figure 3 gives the relaxation temperature the (taken from ε'' and $tg\delta$ spectra) of γ,β,α processes respectively versus lg ω . The relaxation temperatures of all processes increase with rising frequency. Two peaks observed for the δ transition associated with the dipole motion in the crystalline nematic states. These peaks merge into one peak at the temperature of isotropization. The γ relaxation process visible at frequencies higher Hz.

The dielectric loss factor ϵ'' for β and part of α absorption process versus lg frequency is presented in Figure 4. The curves at low temperatures are broad, suggests a wide variety of constrains on the dipole motion. As the temperature increases the whole system moves faster narrowing the loss peak of the β process. Simultaneously an starts to appear at low frequency. The observed of at low frequencies starts at temperature near 0°C. In this region in the vicinity of T the ionic conductivity σ is very low (see figure 5). A fast increase of ionic conductivity in the vicinity of the δ relaxation process in the low frequences range is visible (as illustrated by a lack of σ dependence on frequency), but does not affect yet the behaviour of the β and α relaxation processes.

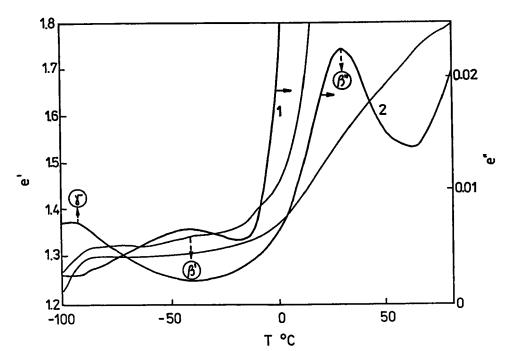


FIGURE 2 ϵ'' and ϵ' spectra for two different frequencies: curve 1-0.03 Hz and curve 2-30000 Hz.Primary β' and secondary β'' relaxations are shown.

Cole-Cole semi-circle curves representing the loss factor ε'' versus ε' are drawn in Figure 6 for the β process. frequency range and at higher temperature relaxation begins to merge with the β relaxation process and Cole-Cole equation does not the experimental data. The relaxation strength $\Delta \varepsilon' = \varepsilon' - \varepsilon'$ associated with reorientation of the dipole moment in the β is relatively small and its increase in is clearly visible. relaxation vicinity of Tg The amplitude ϵ'' rises with increasing temperature. In vicinity of Tg it's jump is observed (see insert Figure 6).

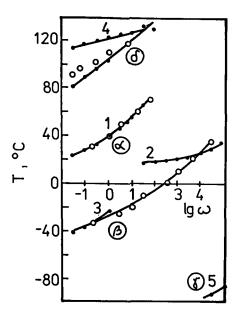


FIGURE 3 The relaxation temperature T versus $\lg \omega$ for γ, β, α and δ processes (open points are taken from $\lg \delta$ versus temperature).

In Fig 7 two examples of fitting of the experimental data obtained at -10°C (curve 1) and 0°C (curve 2) with the common distribution functions of Fuoss-Kirkwood (equation (1)) and of Cole-Cole (equation (2)) are given

$$\cosh^{-1}(\epsilon''_{\text{max}}/\epsilon'') = \beta \ln \omega \tau \tag{1}$$

$$\epsilon'_{\text{o}} - \epsilon'_{\text{max}} = 2\epsilon''_{\text{max}}/\beta$$

$$\epsilon'' = \frac{\frac{1}{2}(\epsilon'_{0} - \epsilon'_{0}) \cos \alpha \pi / 2}{\cosh (1 - \alpha) S + \sin \alpha \pi / 2} \epsilon'' = (\epsilon'_{0} - \epsilon_{\infty})^{1 - \alpha}$$
(2)

where: β - Fuoss-Kirkwood width parameter

 α - Cole-Cole width parameter

γ - additional skewing parameter

 $S - lg(\omega/\omega_{max})$

Curve 5 fits well the experimental results with one fitting parameter $\beta=0.47$ or two fitting parameters $\alpha=0.4$ and $\gamma=0.415$ Curves 3 and 4 are fitted with parameters $\beta=0.52$ and $\beta=1$ (description of the experimental results obtained in low frequency measurements).

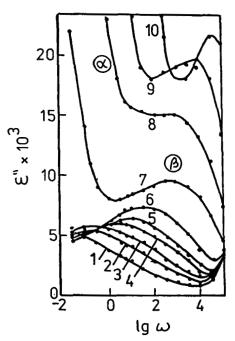


FIGURE 4 The dielectric loss ϵ' as a function of $\lg \omega$ drawn for different temperature from-50 to 35°C in the region of β and α relaxation processes.

Cu r ves 1-10 represents temperatures -50°, -40°, -33°, -25°, -20°, -10°, 0°, 10°, 20° and 35°C.

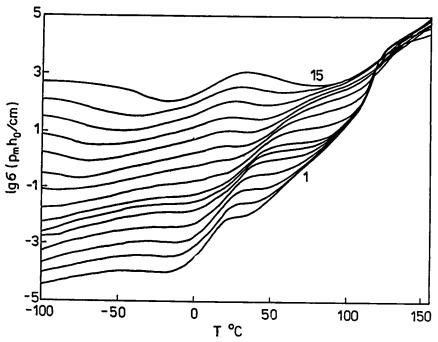


FIGURE 5 lg ionic conductivity σ versus temperature. Curves 1-15 are taken for different frequencies and cover the range from 0.03 Hz (curve 1) to 1000000 Hz (curve 15).

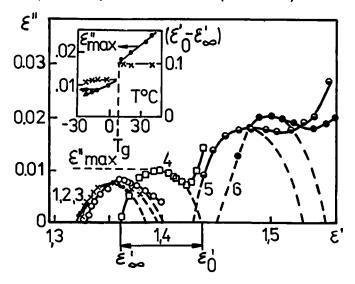


FIGURE 6 Cole-Cole plots for temperature range from -21 to 20°C.Inside: ϵ " and ϵ '_- ϵ '_ versus T.

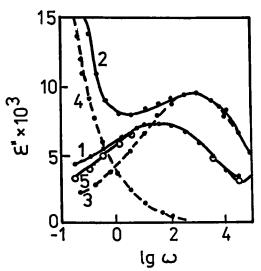


FIGURE 7 ϵ' versus $\lg \omega$. Curves 1- and 2-experimental; curves 3- and 4- Fuoss-Kirkwood distribution curves with β =1 and β =0.52; curve 5- Fuoss-Kirkwood and Cole-Cole distribution curves with β =0.47, α =0.415.

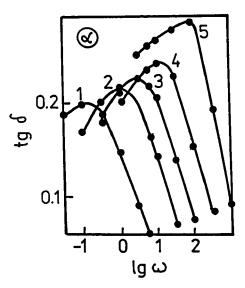


FIGURE 8 tg δ versus lg ω in α process region. Curves 1-5 represent temperatures 30°,40°,50°,60 and 70°C.

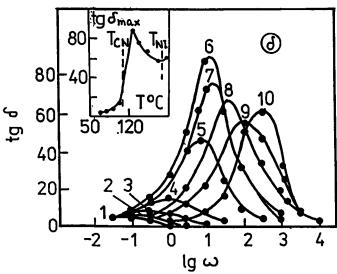


FIGURE 9 tg δ versus lg ω in the δ process region. Curves 1-10 represent temperatures 92°,97°,102°,110°, 115°,125°,130°,140°,150°,160°C. Iside:tg δ _{max} vs. T

Figures 8 and 9 represent the dielectric loss tg δ versus lg ω for the α relaxation process (temperature range 30°-70°C -curves 1-5) and the δ relaxation process (temperature range 92-160°curves 1-10) respectively. The shift of tg δ max higher values in the region of the crystalline-nematic transition is due to the increase in dipole alignment. The subsequent drop of Tg δ is associated with an increase in thermal motion of the whole system on izotropisation and presence of ionic conductivity (see insert Figure 9).

of the relaxation frequencies versus of the lg reciprocal temperature fitted to an Arrhenius relation are presented in Figures 10 and 11. In the case relaxation process a strong devation Arrhenius from behaviour was expected1,3.Also, the activation energy value of the α process was expected to be higher than that of the ß process. However at hight enough frequencies the spectra of both processes (associated with the motion of the could merge together at higher temperatures, dipoles)

leading to similar values of their activation energies. The values of the apparent activation energies associated with these processes are: $E(\beta)=50kJ/mol$, $E(\alpha)=48kJ/mol$, $E(\delta)=75kJ/mol$ (in the solid state) and $E(\delta)=43kJ/mol$ (in the nematic state).

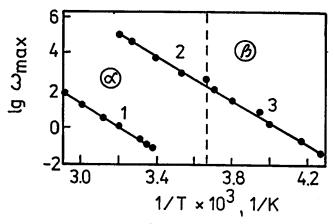


FIGURE 10 lg ω versus reciprocal temperature. Plot 1- α relaxation process, plots 2- and 3- β relaxation processes (primary and secondary).

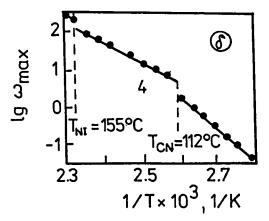


FIGURE 11 $\lg \omega_{\max}$ versus reciprocal temperature. Plot 4-the δ relaxation process.

CONCLUSIONS

Dielectric measurements are a powerfull tool in the study of molecular dynamics of liquid-crystalline polymers. Dielectric relaxation processes γ, β, α and δ can be clearly observed. Cooperative motion in the vicinity of Tg leads to a combined $\alpha+\beta$ relaxation process. Discontinuous change of dielectric strength $\Delta \epsilon'$ at Tg is observed. The δ process is visible in the vicinity of phase transitions of the polymer. of the β,α δ processes were Activation energies and in the value of E for the δ process calculated. The drop from 75 to 43 kJ/mol at the solid - nematic transition connected with the difference in the dynamics of longitudinal component of the dipole moment in these phases.

ACKNOWLEDGEMENT

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